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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FG1386). Services for accessing these data are described at the back of the journal.

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Tetrakis[1-(1-phenylcyclohexyl)piperidinium] Tetrachloromanganate(II) Dichloride

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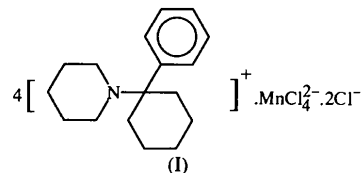
Abstract

The crystal structure of the title compound, $(C_{17}H_{26}N)_4[MnCl_4]Cl_2$, has been determined. The cation is 1-(1-phenylcyclohexyl)piperidinium. The metal atom is located on a $\bar{4}$ rotation axis with four Cl atoms coordinated to form a flattened tetrahedron.

† Deceased.

Comment

The present structural investigation was carried out because of our interest in the geometry and environment of the tetrachlorometallate(II) anion (Harlow *et al.*, 1974, 1975; Nelson & Simonsen, 1981). The structure of the title compound, (I), is isostructural with the Cu



and Ni salts. The four Cl atoms around each metal atom form a flattened tetrahedron with two equal large angles and four equal smaller angles: 121.10(3) and 103.99(1)° for Mn, 132.6(1) and 99.3(1)° for Cu, and 122.4(1) and 103.4(1)° for Ni (Nelson & Simonsen, 1981). Each Cl⁻ ion is hydrogen bonded to the N atoms of two symmetry-related 1-(1-phenylcyclohexyl)-piperidinium (PCP) cations [N...Cl 3.183(2) Å].

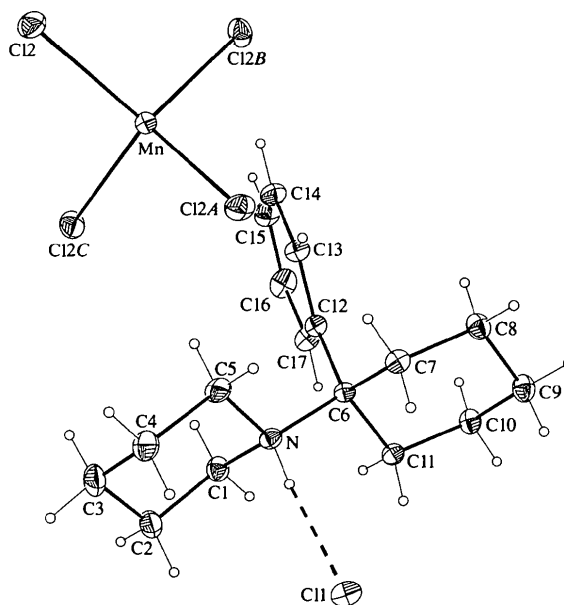


Fig. 1. A view of the title compound showing the atom-labeling scheme. Displacement ellipsoids are scaled to the 30% probability level. H atoms are drawn to an arbitrary scale.

Experimental

Crystals of the title compound were provided by Dr William J. Wells III.

Crystal data

$(C_{17}H_{26}N)_4[MnCl_4]Cl_2$
 $M_r = 1245.20$

Mo $K\alpha$ radiation
 $\lambda = 0.71073 \text{ \AA}$

Tetragonal

*P*4₂*c**a* = 15.747 (1) Å*c* = 13.148 (1) Å*V* = 3260.3 (4) Å³*Z* = 2*D*_x = 1.268 Mg m⁻³*D*_m not measured

Cell parameters from 35 reflections

 $\theta = 24.7\text{--}25.0^\circ$ $\mu = 0.492\text{ mm}^{-1}$ *T* = 173 (2) K

Prism

0.80 × 0.70 × 0.46 mm

Colorless

SHELXL93. Software used to prepare material for publication: *SHELXL93*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1313). Services for accessing these data are described at the back of the journal.

Data collection

Siemens diffractometer

 ω scan

Absorption correction: none

12 559 measured reflections

3237 independent reflections

2777 reflections with

 $I > 2\sigma(I)$ *R*_{int} = 0.046 $\theta_{\text{max}} = 32.50^\circ$ *h* = -23 → 23*k* = -23 → 23*l* = 0 → 19

4 standard reflections

every 96 reflections

intensity decay: 2%

Refinement

Refinement on *F*² $R[F^2 > 2\sigma(F^2)] = 0.035$ $wR(F^2) = 0.089$ *S* = 1.06

3237 reflections

181 parameters

H atoms not refined

 $w = 1/[\sigma^2(F_o^2) + (0.0448P)^2 + 0.5476P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} = -0.04$ $\Delta\rho_{\text{max}} = 0.29\text{ e \AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.27\text{ e \AA}^{-3}$

Extinction correction:

SHELXL93 (Sheldrick, 1993)

Extinction coefficient:

0.0125 (10)

Scattering factors from

International Tables for Crystallography (Vol. C)

Absolute structure: Flack

(1983)

Flack parameter = 0.00 (3)

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Acta Cryst. (1998). **C54**, 476–479Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U^{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
Mn	0	0	0	0.02046 (12)
Cl2	-0.11804 (3)	-0.05617 (4)	0.08839 (5)	0.03157 (13)
C1	0.34898 (15)	-0.09177 (14)	-0.2155 (2)	0.0273 (4)
C2	0.3723 (2)	-0.17418 (15)	-0.1616 (2)	0.0353 (5)
C3	0.3048 (2)	-0.1993 (2)	-0.0846 (2)	0.0415 (6)
C4	0.2916 (2)	-0.1269 (2)	-0.0101 (2)	0.0366 (5)
C5	0.26981 (14)	-0.04437 (15)	-0.0645 (2)	0.0293 (5)
C6	0.32578 (12)	0.06796 (13)	-0.19102 (14)	0.0193 (3)
C7	0.31490 (13)	0.13391 (13)	-0.10540 (15)	0.0233 (4)
C8	0.31498 (15)	0.22439 (14)	-0.1483 (2)	0.0290 (4)
C9	0.3960 (2)	0.24432 (15)	-0.2057 (2)	0.0314 (5)
C10	0.41190 (15)	0.17850 (14)	-0.2887 (2)	0.0266 (4)
C11	0.40945 (13)	0.08783 (14)	-0.2461 (2)	0.0221 (3)
C12	0.24903 (13)	0.06421 (13)	-0.26199 (14)	0.0220 (4)
C13	0.16673 (13)	0.07913 (15)	-0.2259 (2)	0.0271 (4)
C14	0.0971 (2)	0.0767 (2)	-0.2902 (2)	0.0342 (5)
C15	0.1082 (2)	0.0584 (2)	-0.3923 (2)	0.0383 (6)
C16	0.1888 (2)	0.0417 (2)	-0.4294 (2)	0.0384 (6)
C17	0.25884 (15)	0.0451 (2)	-0.3658 (2)	0.0294 (4)
Cl1	0	-1/2	-0.00298 (5)	0.02837 (14)
N	0.33815 (10)	-0.02079 (10)	-0.13987 (13)	0.0213 (3)

The labels of the *x* and *y* axes were chosen to minimize the absolute Flack parameter (Sheldrick *et al.*, 1985).

Data collection: *XSCANS* (Siemens, 1994). Cell refinement: *XSCANS*. Data reduction: *XSCANS*. Program(s) used to solve structure: *SHELXL93* (Sheldrick, 1993). Program(s) used to refine structure: *SHELXL93*. Molecular graphics:

Diiodotetrakis[tris(dimethylamino)-phosphine oxide-*O*]bismuth(III) Pentaiodide, [Bi₂{OP(NMe₂)₃}₄][I₅]

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Abstract

The title compound, [Bi₂(C₆H₁₈N₃OP)₄](I₅), contains the pseudo-octahedral Bi^{III} cation [Bi₂{OP(NMe₂)₃}₄]⁺, which has crystallographic $\bar{4}$ symmetry, and *trans* iodide ligands. The central I atom of the disordered [I₅]⁻ anion is also situated on a site of $\bar{4}$ symmetry and the atoms of this anion form an infinite three-dimensional network, resulting in a 'cage'-type structure. The cations are situated within the 'cage' cavities.